



Pergamon

Tetrahedron Letters 41 (2000) 7383–7386

TETRAHEDRON
LETTERS

A synthesis of trifluoromethyl-substituted naphthalenes

John M. Mellor,^{a,*} Afaf H. El-Sagheer^a and Ezz El-Din M. Salem^b

^a*Department of Chemistry, University of Southampton, Southampton SO17 1BJ, UK*

^b*Department of Chemistry, Faculty of Science, Suez Canal University, Ismailia, Egypt*

Received 9 June 2000; revised 14 July 2000; accepted 20 July 2000

Abstract

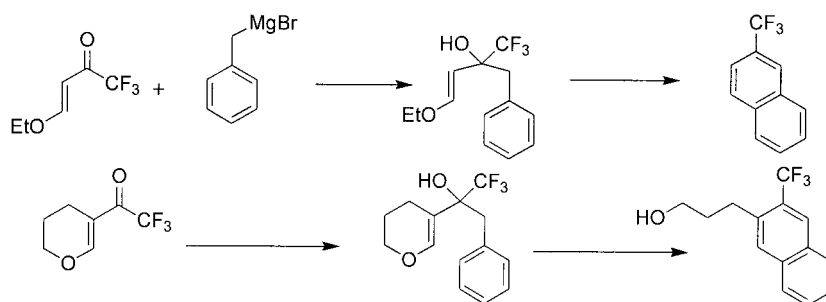
Alkyl and aryl Grignard reagents add to 1-(3,4-dihydro-2*H*-5-pyranyl)-2,2,2-trifluoro-1-ethanone by 1,4-addition, but benzyl Grignard reagents react in good yield by 1,2-addition. Dehydration of the resulting alcohols affords intermediate dienes, which readily undergo cyclisation to give substituted trifluoromethyl-naphthalenes. Addition to other trifluoromethylketones permits access to a range of novel fluorinated naphthalenes and benzenes. © 2000 Published by Elsevier Science Ltd.

Keywords: additions; Grignard reagents; naphthalenes; trifluoromethylketones.

Acyclic fluorinated building blocks¹ have been used widely in the synthesis of fluorinated heterocyclic compounds. Introduction of trifluoromethyl side chain substituents into aromatic carbocycles via a building block approach permitting construction of the ring with incorporation of the trifluoromethyl substituent is little developed. The possibility that the [3+3] benzannulation procedures, recently developed by Junjappa et al.² and reviewed by Katritzky et al.,³ might be used to give fluorinated naphthalenes, is attractive. In this methodology an unsaturated ketone is reacted with a benzyl organometallic reagent to afford a 1,2-adduct, which on dehydration affords aromatic products. In order that this procedure might be applied to trifluoromethylketones, thus giving trifluoromethyl substituted aromatic compounds, it is essential that the benzyl organometallic reagent undergo 1,2-addition with, for example, a β -alkoxy- α,β -unsaturated ketone, rather than the possible 1,4-addition (see Scheme 1).

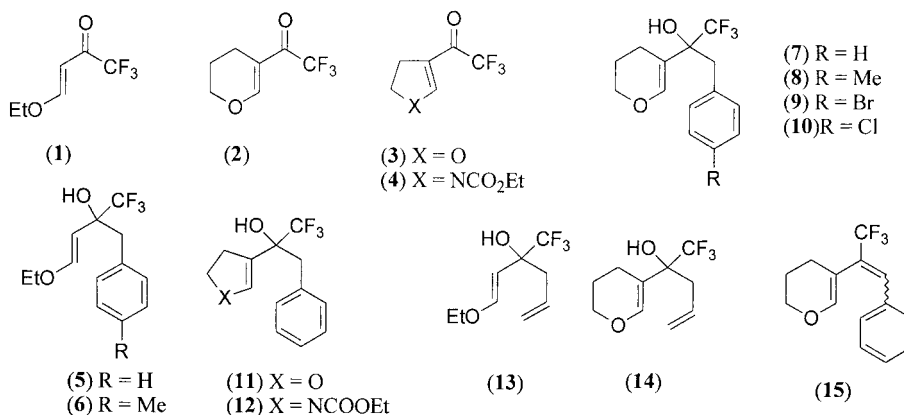
In the case of additions of Grignard reagents to unsaturated ketones, not having β -alkoxy- or other heteroatom substitutions, it has been recognised⁴ that a charge control preferentially leads to 1,2-addition, but, in contrast, an orbital control leads preferentially to 1,4-addition. Allyl Grignards uniformly react⁵ preferentially by 1,2-addition, but examples are known⁶ of preferential 1,4-addition of benzyl Grignard reagents and in many examples⁷ alkyl, aryl and benzyl Grignard reagents give mixtures of products. Examples have been reported by Gustafsson,⁸ where benzyl magnesium halides preferentially react by 1,2- or 1,4- depending upon the nature of the halide. In the case of β -alkoxy- α,β -unsaturated methyl ketones⁹ reaction with

* Corresponding author. E-mail: jmm4@soton.ac.uk



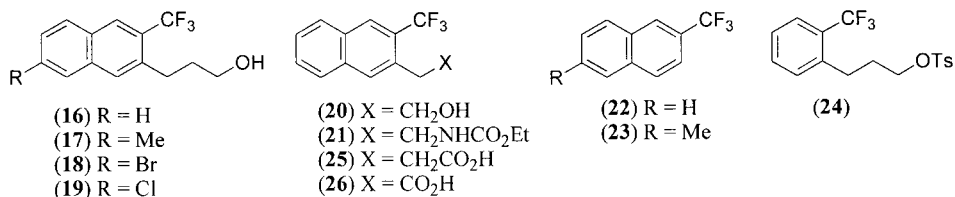
Scheme 1.

phenyl magnesium bromide gives both 1,2- and 1,4-adducts. Cases having both a β -alkoxy-substituent and trifluoromethyl-substitution have been little examined. With the β -alkoxy- α,β -unsaturated trifluoromethyl ketone (**1**)¹⁰ aryl magnesium bromides undergo 1,4-addition. We establish in this paper that, in contrast to alkyl and aryl Grignard reagents, which preferentially undergo 1,4-addition, reaction with benzyl Grignard reagents occurs by 1,2-addition, thus enabling us to synthesise a range of substituted trifluoromethylnaphthalenes.



The required acyclic and cyclic β -alkoxy- α,β -unsaturated trifluoromethyl ketones (**1–3**)¹¹ are readily available and a β -amino analogue **4**,¹² prepared from proline, has recently been described. Reactions of both the acyclic ketone **1** and the cyclic ketone **2** with phenyl magnesium bromide and alkyl magnesium bromides give mainly products of 1,4-addition; e.g. phenyl magnesium bromide and octyl magnesium bromide give by 1,4-addition with the ketone **2** *cis*-products in 79 and 67% yields, respectively. Similarly, ketone **1** reacts¹⁰ with phenyl magnesium bromide via 1,4-addition, but not 1,2-addition. In sharp contrast we find that benzyl magnesium bromides give 1,2-adducts with the cyclic ketone **2** in satisfactory yields¹³ (see Table 1). The structures of the 1,2-adducts are established by observation of the signal of the vinylic proton at about δ 6.5 ppm and the OH absorption at 3450 cm^{-1} . A structural confirmation was obtained¹⁴ by a single crystal X-ray analysis of the adduct **9**, prepared from ketone **2**. Similarly, benzyl magnesium halides preferentially undergo 1,2-addition to the acyclic ketone **1**, the cyclic ketone **3** and the pyrrole analogue **4** affording the alcohols (**5**, **6**, **11** and **12**). The spectroscopic data obtained for all the alcohols (**5–12**) reported in Table 1 are in good agreement with data obtained for the alcohol **9**. These additions establish that a route to trifluoromethylnaphthalenes can be based on

Table 1



Ketone	Grignard Reagent	Alcohol	(Yield %)	Aromatic (Yield %)
(1)	PhCH ₂ MgBr	(5)	55	(22) 84
(1)	<i>p</i> -MeC ₆ H ₄ CH ₂ MgBr	(6)	45	(23) 95
(2)	PhCH ₂ MgBr	(7)	93	(16) 82
(2)	<i>p</i> -MeC ₆ H ₄ CH ₂ MgBr	(8)	97	(17) 90
(2)	<i>p</i> -BrC ₆ H ₄ CH ₂ MgBr	(9)	74	(18) 92
(2)	<i>p</i> -ClC ₆ H ₄ CH ₂ MgBr	(10)	80	(19) 94
(3)	PhCH ₂ MgBr	(11)	34	(20) 92
(4)	PhCH ₂ MgBr	(12)	25	(21) 63
(2)	CH ₂ CH=CHMgBr	(14)	82	(24) 60

a preferred 1,2-addition of benzyl Grignards to cyclic α,β -unsaturated trifluoromethyl ketones, having β -heteroatom substituents. Allyl magnesium bromide also adds to the ketones **1** and **2** by 1,2-addition, giving the alcohols **13** and **14** and crotyl magnesium bromide and cinnamyl magnesium bromide add to the ketones **1** and **2** by 1,2-addition.

An effective procedure for dehydration affording naphthalenes was established using *p*-toluenesulphonic acid in toluene as the catalyst. Under mild conditions dehydration of adduct **7** gave the dienes **15** in 84% yield. Under more vigorous conditions the adduct **7** gave the naphthalene **16** in 82% yield. Under similar conditions the dienes **15** gave the naphthalene **16**. Adducts from the cyclic ketones **2** and **3** gave the alcohols (**17**–**20**) in good yield (see Table 1). Similarly, dehydration of alcohol **12** gave the naphthalene **21** having an amide side-chain in 63% yield. The procedure is also effective with adducts derived from the acyclic ketone **1**. Dehydration of the adducts **5** and **6** afforded the known 2-trifluoromethylnaphthalene¹⁵ **22** and 2-methyl-6-trifluoromethylnaphthalene **23**, respectively. Reaction of the adduct **14** under similar conditions, but for a longer period, gave the tosylate **24**.

The major significance of these results is the development of a regiospecific route to 2-trifluoromethyl benzenes and naphthalenes. 3,6-Disubstituted 2-trifluoromethyl-naphthalenes have not been prepared prior to our work. The interest in the side-chain alcohols is enhanced by their possible transformation to other substituents, thus offering a wider range of aromatic end-groups. The point is illustrated by oxidation using Jones' reagent of the alcohols **16** and **20** to the naphthyl-propionic and acetic acids **25** (90%) and **26** (85%), respectively. 3-Substituted-2-trifluoromethylnaphthalenes have already been prepared¹⁶ in a study of candidate drugs to be used against diabetes. Our access to a series of trifluoromethyl-substituted naphthalenes carrying additional alcohol, acid and amine end-groups promises much scope for structural elaboration.

Acknowledgements

We thank the Egyptian Government for a studentship (A.H.El-S.) and EPSRC for X-ray facilities.

References

1. Tozer, M. J.; Herpin, T. F. *Tetrahedron* **1996**, *52*, 8619; Percy, J. M. *Contemp. Org. Synth.* **1995**, *2*, 251.
2. Junjappa, H.; Ila, H.; Asokan, C. V. *Tetrahedron* **1990**, *46*, 5423.
3. Katritzky, A. R.; Li, J.; Xie, L. *Tetrahedron* **1999**, *55*, 8263.
4. Cossentini, M.; Deschamps, B.; Anh, N. T.; Seyden-Penne, J. *Tetrahedron* **1977**, *33*, 409.
5. El Idrissi, M.; Santelli, M. *J. Org. Chem.* **1988**, *53*, 1010.
6. Gillespie, J. S.; Acharya, S. P.; Shamblee, D. A.; Davis, R. E. *Tetrahedron* **1975**, *31*, 3.
7. Imamoto, T.; Takiyama, N.; Nakamura, K.; Hatajima, T.; Kamiya, Y. *J. Am. Chem. Soc.* **1989**, *111*, 4392; House, H. O.; Traficante, D. D.; Evans, R. A. *J. Org. Chem.* **1963**, *28*, 348; Holm, T. *Acta Chem. Scand.* **1991**, *45*, 925.
8. Gustafsson, B. *Acta Chem. Scand.* **1977**, *31B*, 382.
9. Spassky-Pasteur, A.; Riviere, H. *Bull. Soc. Chim. France* **1969**, 811.
10. Gorbunova, M. G.; Gerus, I. I.; Kukhar, V. P. *J. Fluorine Chem.* **1993**, *65*, 25.
11. Hojo, M.; Masuda, R.; Sakaguchi, S.; Takagawa, M. *Synthesis* **1986**, 1016; Buback, M.; Tost, W.; Hubsch, T.; Vob, E.; Tietze, L. F. *Chem. Ber.* **1989**, *122*, 1179.
12. Kawase, M.; Hirabayashi, M.; Koiwai, H.; Yamamoto, K.; Miyamae, H. *Chem. Commun.* **1998**, 641; Oliveira, D. F.; Miranda, P. C. M. L.; Correia, C. R. D. *J. Org. Chem.* **1999**, *64*, 6646.
13. All new compounds were characterised spectroscopically and by high resolution mass spectroscopy, or by combustion analysis.
14. Coles, S.; Hursthouse, M. B. personal communication.
15. Matthews, D. P.; Whitten, J. P.; McCarthy, J. R. *Tetrahedron Lett.* **1986**, *27*, 4861.
16. Ellingboe, J. W.; Lombardo, L. J.; Alessi, T. R.; Nguyen, T. T.; Guzzo, F.; Guinasso, C. J.; Bullington, J.; Browne, E. N. C.; Bagli, J. F.; Wrenn, J.; Steiner, K.; McCaleb, M. L. *J. Med. Chem.* **1993**, *36*, 2485.